nannvi

Asymmetric Direct Vinylogous Michael Addition to 2‑Enoylpyridine N‑Oxides Catalyzed by Bifunctional Thio-Urea

Subhrajit Rout,[†] Sumit K. Ray,[‡] Rajshekhar A. Unhale,[†] and Vinod K. Singh^{*,†,‡}

† Department of Chemistry, Indian Institute of Science Education and Research Bhopal, Bhopal [- 4](#page-2-0)62 066, MP, India ‡ Department of Chemistry, Indian Institute of Technology Kanpur, Kanpur - 208 016, UP, India

S Supporting Information

ABSTRACT: Catalytic enantioselective direct vinylogous Michael addition of α,α -dicyanoalkenes to 2-enoylpyridine N-oxides with a bifunctional organocatalyst is described. The methodology offers an efficient way to install an asymmetric carbon−carbon bond at the γ -position of $\alpha_i \alpha$ -dicyanoalkenes in excellent regio-, diastereo-, and enantioselectivity. Further, application in desymmetrization of achiral α,α-dicyanoalkene to access highly functionalized enantioenriched cyclohexylidenemalononitrile derivatives has been demonstrated.

 \sum yridine and their N-oxides are ubiquitous structural motifs
in a wide range of bioactive natural products with remarkable biological properties.¹ These heterocycles prevail in many pharmaceuticals, agrochemicals, and functional m aterials² and constitute importa[nt](#page-3-0) scaffolds in natural product synthesis. 3 In addition, enantioenriched pyridine N-oxides have also gain[ed](#page-3-0) greater attention in the field of asymmetric catalysis and eme[rg](#page-3-0)ed as an efficient organocatalyst as well as ligand.^{4,5} In this context, 2-enoylpyridine N-oxide has appeared as an efficient and privileged substrate for various asymmet[ric](#page-3-0) transformations.⁶ In the past decade, we and others have reported asymmetric Friedel−Crafts, Michael, and Mukaiyama−Michael r[ea](#page-3-0)ctions using 2-enoylpyridine N-oxide as a template in the presence of catalytic metal complexes.⁷ However, no report has been disclosed utilizing 2-enoylpyridine N-oxide as a substrate under asymmetric organocatalysis, to th[e](#page-3-0) best of our knowledge.

Asymmetric direct Michael reaction under vinylogous nucleophilicity offers a powerful way to access enantioenriched γ-functionalized α , β -unsaturated carbonyl compounds with the introduction of the concept of vinylogy. The strategy has gained much attention from the atom economy viewpoint as well as green chemistry. Among various synthetic protocols of the asymmetric direct vinylogous Michael reaction, addition of α , α -dicyanoalkenes to different electrophilic partners provides enantioenriched γ-carbon-functionalized exocyclic or acyclic activated alkenes.⁸ This has been nicely explored by Deng et al. on α , β -unsaturated ketones⁹ and by Chen et al. with N-Boc aldimines as elec[tr](#page-3-0)ophilic partners¹⁰ in their pioneering reports.

In the past decade, thio-urea (urea) catalysts with the $cinchona$ motif 11 have appreciably influenced many asymmetric transformations highlighting enhanced catalytic ability via activation of [bo](#page-3-0)th partners in a biomolecular reaction and a high level of stereodifferentiation under mild reaction conditions.¹² Inspired by such catalytic activity, herein, we report the first catalytic asymmetric and direct vinylogous Michael re[ac](#page-3-0)tion of α , α -dicyanoalkenes to 2-enoylpyridine Noxides with cinchona derived bifunctional thio(urea) as an organocatalyst.

At the outset, a model reaction comprising vinylogous donor 2a and Michael acceptor 3a was carried out using 5 mol % of quinine derived urea catalyst 1a in toluene at rt. To our delight, the corresponding vinylogous Michael product was found in exclusive γ-selectivity in high yield with excellent diastereo- and enantioselectivity (Table 1, entry 1). With this result, a variety of *cinchona-*derived bifunctional organocatalysts 1a−j¹³ (Figure 1) were then screened. [I](#page-1-0)t was found that, in most cases, products were obtained in good to excellent enanti[ose](#page-3-0)lectivity [\(T](#page-1-0)able 1). The best result was achieved in the case of cinchonidine derived thiourea 1f, in terms of enantioselectivity (Table [1,](#page-1-0) entry 6). The opposite enantiomer of the Michael product could be accessed by using pseudoenantiomeric catalyst[s.](#page-1-0) Complete optimization studies (entries 11−14) revealed that the best result could be achieved in the case of 10 mol % 1f (Table 1, entry 11).

Received: September [1](#page-1-0), 2014 Published: October 13, 2014

Table 1. Optimization of Direct Vinylogous Michael Reaction^{a, \boldsymbol{b}}

^aReactions were carried out on 0.1 mmol of 3a and 0.13 mmol of 2a in 1.0 mL of toluene at rt, unless otherwise noted. $\frac{b}{c}$ dr in all cases is found to the >99:1 from ¹H NMR analysis. ^cIsolated yield. ^{*d*} Determined by HPLC using chiral IA-3 column. ^e Opposite enantiomer as major was obtained.

Figure 1. Cinchona alkaloid derived thio-urea (urea) catalysts.

Subsequently, a series of solvents were screened at rt in order to investigate the effect on enantioselectivity in different solvents (Table 2). It was found that most of the solvents worked well for vinylogous Michael addition to offer the product in 90−95% ee (Table 2, entries 1−7). In the case of acetonitrile, the product was found in 94% yield and 78% ee (Table 2, entry 8). It was also observed that both toluene and 1,4-dioxane offered the vinylogous Michael product in 95% ee (Table 2, entries 1 and 7). However, toluene was chosen for further studies from the reaction time standpoint. Further, lowering the reaction temperature from rt to 0 °C, we obtained the product with enhanced enantioselectivity (Table 2, entry 9). No change in enantioselectivity was observed by decreasing the temperature from 0 $^{\circ}$ C to -25 $^{\circ}$ C; rather the reaction slowed down (Table 2, entry 10). Thus, we chose 10 mol % catalyst 1f in toluene at 0° C for further studies.

Next, the substrate scope for direct vinylogous Michael addition of a variety of α , α -dicyanoalkenes 2a−g (Figure 2) to a

Table 2. Solvent Study under Different Temperature^{a,b}

	NC. .CN $\ddot{}$ Phi 2a	$\frac{0}{N}$ 3a	1f (10 mol %) solvent, temp	NC. .CN Å ${}^{\!\!\!\!\!\!\sim}$ Ph 4aa	
entry	solvent	temp $(^{\circ}C)$	time (h)	yield ^c $(\%)$	ee ^d $(\%)$
$\mathbf{1}$	toluene	rt	24	95	95
2	p -xylene	rt	26	91	93
3	CH_2Cl_2	rt	31	88	90
4	DCE	rt	29	88	90
5	CHCl ₃	rt	31	89	94
6	THF	rt	30	90	94
7	1,4-dioxane	rt	28	91	95
8	CH ₃ CN	rt	28	94	78
9	toluene	Ω	30	95	98
10	toluene	-25	60	94	98

^aReactions were carried out on 0.1 mmol of 3a and 0.13 mmol of 2a in 1.0 mL of solvent at rt, unless otherwise noted. ^bdr in all cases is found to the >99:1 from ¹H NMR analysis. ^cIsolated yield. ^dDetermined by HPLC using chiral IA-3 column.

Figure 2. Different α , α -dicyanoalkenes studied.

variety of 2-enoylpyridine N-oxides 3a−l was studied. It is worthwhile to note that the protocol worked well for both aryland heteroaryl-substituted 2-enoylpyridine N-oxides to achieve corresponding adducts with high enantioselectivity (Table 3, entries 1−9). Later, 2-enoylpyridine N-oxide bearing a cinnamyl group 3j offered the product in 88% ee (Table [3,](#page-2-0) entry 10). However, ethyl glyoxalate derived 2-enoylpyridine Noxide 3k yielded the product with a 90% yield and 60% [ee](#page-2-0) (Table 3, entry 11). The α,α-dicyanoalkenes 2b−d reacted smoothly to afford the products in the range of 90−96% ee (Table [3,](#page-2-0) entries 12−16).

Later, acyclic α,α-dicyanoalkenes 2e−f were also employed under [op](#page-2-0)timized conditions and it was found that these compounds also worked well in the asymmetric direct vinylogous Michael addition leading to high enantioselectivities, however, with a longer reaction time as compared to 2a−d (Table 3, entries 17−18). The cyclohexanone derived α , α dicyanoalkene also produced the vinylogous product in 84% yield a[nd](#page-2-0) 90% ee (Table 3, entry 19). It is important to mention that no polymerization of α , α -dicyanoalkenes 2a−g was observed as an unwan[ted](#page-2-0) side reaction under optimized reaction conditions.

The single crystal X-ray analysis of vinylogous product 4ca confirmed the absolute stereochemistry to be (9R, 10S) (Figure 3). Eventually, the reaction was carried out in gram scale to highlight the efficiency of the protocol (Scheme 2).

Further, we thought of exploring the scope of our [m](#page-2-0)ethodology to a desymmetrization strate[gy](#page-2-0) using 4-

Table 3. Substrate Scope of Direct Vinylogous Michael Reaction a,b

a Reactions were carried out on 0.2 mmol of 3 and 0.26 mmol of 2 in 1.5 mL of toluene at 0 $^{\circ}$ C, unless otherwise noted. b dr in all cases is Found to be >99:1 from $\frac{1}{1}$ NMR analysis. Tsolated yield.
 $\frac{d}{dt}$ Determined by HPIC using chiral column "Enantioselectivity" Determined by HPLC using chiral column. ^eEnantioselectivity (99.9) was obtained after recrystallization, and absolute stereochemical outcome was determined as (9R, 10S) by X-ray structural analysis.

Figure 3. X-ray structure of 4ca.

substituted cyclohexanone derived achiral α , α -dicyanoalkene. We envisioned this would produce highly functionalized enantioenriched activated methylenecyclohexane derivatives with three chiral centers (Scheme 1) found as core structures of many natural products.¹⁴ Toward this, 4-phenyl cyclohexanone derived achiral $α, α$ -dicyanoalkene 5 was reacted with 2-enoylpyridine N-oxides 3b [a](#page-3-0)nd 3d in the presence of 10 mol % of 1f (Scheme 1). Gratifyingly, this process enabled us to access highly functionalized cyclohexane derivatives in 96−97% ee with excellent dr.

The synthetic utility of the reaction was shown by reduction of vinylogous Michael adduct 4ba with Zn dust in THF in the presence of saturated $NH₄Cl$ (Scheme 2). This one-pot process afforded hydrofuran fused hexahydrocyclopenta $[c]$ chromane 7 with two contiguous stereogenic quaternary carbons adjacent to

Scheme 1. Desymmetrization of Achiral α,α -Dicyanoalkene 5

two contiguous stereogenic tertiary carbons in 73% yield.¹⁵ The structure and absolute stereochemistry of 7 was confirmed unambiguously by X-ray structure analysis.¹⁶ The c[ore](#page-3-0) of hexahydrocyclopenta $[c]$ chromane with multiple chiral centers is found in several natural products.¹⁷

In conclusion, we disclosed the first asymmetric direct vinylogous Michael addition of α , α -dicyanoalkenes to 2enoylpyridine N-oxides with a bifunctional organocatalyst. This process is operationally simple and afforded a variety of vinylogous Michael products in exclusive γ-selectivity with excellent diastereo- and enantioselectivities. The scope of methodology was further diversified to a desymmetrization strategy to furnish enantioenriched highly functionalized activated methylenecyclohexane derivatives. Further, one of the vinylogous products was transformed to a hydrofuran fused hexahydrocyclopenta $\lceil c \rceil$ chromane framework with four contiguous strereogenic carbon centers. Further study on asymmetric organocatalytic strategies under vinylogous nucleophilicity is currently underway in our laboratory.

■ ASSOCIATED CONTENT

6 Supporting Information

General experimental procedures and analytical data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

■ AUTHOR INFORMATION Corresponding Author

*E-mail: vinodks@iitk.ac.in.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

V.K.S. thanks the Department of Science and Technology (DST), Government of India for a research grant through a J. C. Bose fellowship. S.R. thanks the Council of Scientific and Industrial Research (CSIR), New Delhi for a research fellowship. R.A.U. thanks CSIR for S. P. Mukherjee fellowship. We are thankful to Mr. Chandrajeet Mohapatra, IIT Kanpur for assistance with the X-ray structural analysis and Dr. Vishnumaya Bisai, IISER Bhopal for her kind help.

■ DEDICATION

This work is dedicated to Professor Ganesh Pandey on the occasion of his 60th birthday.

■ REFERENCES

(1) (a) Edrada, R. A.; Proksch, P.; Wray, V.; Witte, L.; Müller, W. E. G.; Van Soest, R. W. M. J. Nat. Prod. 1996, 59, 1056−1060. (b) Kobayashi, J.; Hirasawa, Y.; Yoshida, N.; Morita, H. Tetrahedron Lett. 2000, 41, 9069−9073. (c) Jayasinghe, L.; Jayasooriya, C. P.; Hara, N.; Fujimoto, Y. Tetrahedron Lett. 2003, 44, 8769−8771. (d) Rao, K. V.; Kasanah, N.; Wahyuono, S.; Tekwani, B. L.; Schinazi, R. F.; Hamann, M. T. J. Nat. Prod. 2004, 67, 1314−1318. (e) Morita, H.; Ishiuchi, K.; Haganuma, A.; Hoshino, T.; Obara, Y.; Nakahata, N.; Kobayashi, J. Tetrahedron 2005, 61, 1955−1960. (f) Ishiuchi, K.; Kubato, T.; Mikami, Y.; Obara, Y.; Nakahata, N.; Kobayashi, J. Bioorg. Med. Chem. 2007, 15, 413−417. (g) Ishuichi, K.; Kobota, T.; Ishiyama, H.; Hayashi, S.; Shibata, T.; Mori, K.; Obara, Y.; Nakahata, N.; Kobayashi, J. Bioorg. Med. Chem. 2011, 19, 749−753. (h) Tang, Y.; Fu, Y.; Xiong, J.; Li, M.; Ma, G.-L.; Yang, G.-X.; Wei, B.-G.; Zhao, Y.; Zhang, H.-Y.; Hu, J.-F. J. Nat. Prod. 2013, 76, 1475−1484.

(2) (a) Gonzalez-Bello, C.; Castedo, L. Six-Membered Heterocycles: ́ Pyridines in Modern Heterocyclic Chemistry; Alvarez-Builla, J., Vaquero, J. J., Barluenga, J., Eds.; Wiley-VCH: Weinheim, Germany, 2011; pp 1431−1525. (b) Henry, G. D. Tetrahedron 2004, 60, 6043− 6061.

(3) (a) Zhao, L.; Tsukano, C.; Kwon, E.; Takemoto, Y.; Hirama, M. Angew. Chem., Int. Ed. 2013, 52, 1722−1725. (b) Tsukano, C.; Zhao, L.; Takemoto, Y.; Hirama, M. Eur. J. Org. Chem. 2010, 4198−4200.

(4) Malkov, A. V.; Kočovský. Eur. J. Org. Chem. 2007, 29–36.

(5) Katritzky, A. R.; Lagowski, J. M. In Chemistry of the heterocyclic Noxides; Academic Press: 1971.

(6) (a) Barroso, S.; Blay, G.; Pedro, J. R. Org. Lett. 2007, 9, 1983− 1986. (b) Barroso, S.; Blay, G.; Muñ oz, M. C.; Pedro, J. R. Org. Lett. 2011, 13, 402−405. (c) Barroso, S.; Blay, G.; Muñ oz, M. C.; Pedro, J. R. Adv. Synth. Catal. 2009, 351, 107−111. (d) Livieri, A.; Boiocchi, M.; Desimoni, G.; Faita, G. Chem.-Eur. J. 2011, 17, 516-520.

(7) (a) Singh, P. K.; Singh, V. K. Org. Lett. 2010, 12, 80−83. (b) Singh, P. K.; Singh, V. K. Org. Lett. 2008, 10, 4121−4124. (c) Ray, S. K.; Singh, P. K.; Singh, V. K. Org. Lett. 2011, 13, 5812−5815. (d) George, J.; Reddy, B. V. S. Org. Biomol. Chem. 2012, 10, 4731− 4738. (e) Ray, S. K.; Singh, P. K.; Molleti, N.; Singh, V. K. J. Org. Chem. 2012, 77, 8802−8808. (f) Ray, S. K.; Rout, S.; Singh, V. K. Org. Biomol. Chem. 2013, 11, 2412−2416. (g) Rout, S.; Ray, S. K.; Singh, V. K. Org. Biomol. Chem. 2013, 11, 4537−4545. (h) Livieri, A.; Boiocchi, M.; Desimoni, G.; Faita, G. Chem.-Eur. J. 2012, 18, 11662-11668. (i) Desimoni, G.; Faita, G.; Quadrelli, P. Chem. Rev. 2014, 114, 6081− 6129.

(8) (a) Xue, D.; Chen, Y.-C.; Wang, Q.-W.; Cun, L.-F.; Zhu, J.; Deng, J.-G. Org. Lett. 2005, 7, 5293−5296. (b) Poulsen, T. B.; Bell, M.; Jørgensen, K. A. Org. Biomol. Chem. 2006, 4, 63−70. (c) Xie, J.-W.; Yue, L.; Xue, D.; Ma, X.-L.; Chen, Y. C.; Wu, Y.; Zhu, J.; Deng, J.-G. Chem. Commun. 2006, 1563−1565. (d) Jiang, L.; Zheng, H.-T.; Liu, T.-Y.; Yue, L.; Chen, Y.-C. Tetrahedron 2007, 63, 5123−5128. (e) Lu, J.; Liu, F.; Loh, T.-P. Adv. Synth. Catal. 2008, 350, 1781−1784. (f) Lu,

J.; Liu, F.; Zhao, W.-J.; Loh, T.-P. Tetarhedron Lett. 2008, 49, 5389− 5392. (g) Alemán, J.; Jacobsen, C. B.; Frisch, K.; Overgaard, j.; Jørgensen, K. A. Chem. Commun. 2008, 632−634. (h) Lu, J.; Zhou, W.-J.; Liu, F.; Low, T.-P. Adv. Synth. Catal. 2008, 350, 1796−1800. (i) Cui, H.-L.; Chen, Y.-C. Chem. Commun. 2009, 4479−4486. (j) Schneider, C.; Abels, F. Org. Biomol. Chem. 2014, 12, 3531−3543. (9) Xie, J.-W.; Chen, W.; Li, R.; Zeng, M.; Du, W.; Yue, L.; Chen, Y.- C.; Wu, Y.; Zhu, J.; Deng, J.-G. Angew. Chem., Int. Ed. 2007, 46, 389− 392.

(10) Liu, T.-Y.; Cui, H.-L.; Long, J.; Li, B.-J.; Wu, Y.; Ding, L.-S.; Chen, Y.-C. J. Am. Chem. Soc. 2007, 129, 1878−1879.

(11) Song, C. E., Ed. Cinchona Alkaloids in Synthesis and Catalysis; Wiley-VCH: Weinheim, 2009.

(12) (a) Connon, S. J. Chem.—Eur. J. 2006, 12, 5418-5427. (b) Connon, S. J. Chem. Commun. 2008, 2499−2510.

(13) (a) Vakulya, B.; Varga, S.; Csámpai, A.; Soós, T. Org. Lett. 2005, 7, 1967−1969. (b) Liu, T.-Y.; Long, J.; Li, B.-J.; Jiang, L.; Li, R.; Wu, Y.; Ding, L.-S.; Chen, Y.-C. Org. Biomol. Chem. 2006, 4, 2097−2099. (c) Marcelli, T.; Van der Haas, R. N. S.; Van Maarseveen, J. H.; Hiemstra, H. Angew. Chem., Int. Ed. 2006, 45, 929−931. (d) Wang, Y.- F.; Zhang, W.; Luo, S.-P.; Zhang, G.-C.; Xia, A.-B.; Xu, X.-S.; Xu, D.- Qi. Eur. J. Org. Chem. 2010, 4981−4985.

(14) (a) de la Fuente Martin, G.; Mesía, L. R. Phytochemistry 1997, 46, 1087−1090. (b) Feng, F.; Liu, J.-H.; Zhao, S.-X. Phytochemistry 1998, 49, 2557−2559. (c) Shih, F.-Y.; Chen, T.-H.; Lu, M.-C.; Chen, W.-F.; Wen, Z.-H.; Kuo, Y.-H.; Sung, P.-J. Int. J. Mol. Sci. 2013, 14, 21781−21789.

(15) (a) Li, J.-P.; Zhang, Y.-X.; Ji, Y. J. Chin. Chem. Soc. 2008, 55, 390−393. (b) Schipper, D. J.; El-Salfiti, M.; Whipp, C. J.; Fagnou, K. Tetrahedron 2009, 65, 4977−4983.

(16) See Supporting Information for details.

(17) (a) Namikoshi, M.; Nakata, H.; Yamada, H.; Nagai, M.; Saitoh, T. Chem. Pharm. Bull. 1987, 35, 2761−2773. (b) Yang, B. O.; Ke, C.- Q.; He, Z.[-S.; Yang, Y.-p.; Ye, Y.](#page-2-0) Tetrahedron Lett. 2002, 43, 1731− 1733.